**Electrochemistry**

or, anyway,

**PNP, Well Posed**

Bob Eisenberg

Department of Molecular Biophysics and Physiology

Rush University

Chicago IL 60612

USA

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The work reported here was stimulated by intensive discussions with Chun Liu and YunKyong Hyon, Pennsylvania State University, and would not have occurred without them. The particular formulation of the Poisson Nernst Planck system I use here is that being used in work in collaboration with Allen Tzyy-Leng Horng, Feng Chia University, Taichung, Taiwan; Tai-Chia Lin 林太家, National Taiwan University, Taipei; Xiaofin Lin and his student Allen Flavel, Illinois Institute of Technology, and Chun. Yoichiro Mori (University of Minnesota) has corrected significant errors. Many thanks. Those that remain are of course only my responsibility!

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 Biology and chemistry occur in ionic solutions for the most part. Water without ions is lethal for cells and most proteins. Chemistry is often done in ionic solutions. There seems no need here to repeat the extensive documentation in ([Eisenberg 2012](#_ENREF_10); [Eisenberg 2011a](#_ENREF_7); [Eisenberg 2011b](#_ENREF_8); [Eisenberg 2011c](#_ENREF_9), [2010](#_ENREF_6)) and the historical references to PNP in ([Bazant et al. 2004](#_ENREF_1); [Coalson and Kurnikova 2005](#_ENREF_4); [Damocles 2007](#_ENREF_5); [Eisenberg and Chen 1993](#_ENREF_11); [Eisenberg 1996b](#_ENREF_13), [a](#_ENREF_12); [Jerome 1995](#_ENREF_24); [Selberherr 1984](#_ENREF_34); [Zheng and Wei 2011](#_ENREF_39)) .

 Ions in water like Na+, K+ and Cl− move approximately as hard spheres in a frictional dielectric as described by the implicit solvent (so-called) primitive (or implicit solvent) model of electrolyte solutions. A good representation of this primitive model is given by the Poisson Nernst Planck equations once it is modified to deal with finite size ions.

 Partial differential equations of this sort couple diffusion, migration, and the electric field and form a system that must be mathematically well defined if they are to be solved. The formulation of a well posed version of these equations, starting at time zero, progressing to time infinity, including multiple species is addressed here. I am unaware of previous treatments of this problem that produce well posed mathematical problems for all mixtures of ions on this complete time scale, although such treatments may well exist beyond the horizon, of my knowledge and experience.

 The fundamental difficulty is that the equations have usually been viewed as typical flow, conservation of mass, etc. equations, in the tradition of the fluid mechanics of (mostly uncharged) systems. But the ions flowing are charged, the experiments are designed to deal with electrical current and potential first, and the flux of the ions is not dealt with directly in most experiments. The flux of electrical charge (i.e., the electrical current density) is ***not at all*** the same as the flux of ions or the flux of mass.

**Plan of Attack**

It is my view that the equations of PNP must be written and solved so that ***the electrical properties are dealt with first, and as accurately as possible,*** and residual difficulties, if any, appear in the flux and concentration of the least important ions. Because electrical forces and flows are on a different scale from diffusion forces and flows, this can make a very large difference numerically. In particular, the net charge  must never be computed explicitly. The difference in concentrations of positive and negative charge is well within the round-off error of even double precision computer arithmetic because most of the system is nearly electrically neutral with Errors in computing  can dominate the problem creating dreadful artifacts.

**Shunt Capacitance.** Electrical forces and flows are very large, and act on a very fast time scale (sometimes faster than microseconds). That is one of the issues we deal with here. We include the shunt capacitance always present in the real world, so charge can flow in the equations proportional to  as it does in the real world, when voltages change rapidly.

The capacitive shunt path (typically 100 pF) allows tiny imbalances in  to dissipate very quickly on the capacitive time scale (typically microseconds or faster) without grave effects on the processes that we are interested in, that occur on the time scale of milliseconds. Tiny imbalances in  are hard to avoid, no matter how we try, somewhere in the model, modeling, and computational process. They are likely to be present if initial conditions are set only for the concentrations  because  is so badly determined by  Even if an initial condition is provided for , and for only *N − 1* concentrations, even if , all sorts of irrelevant charging transients are likely to occur. (Consider what happens between  and  when a voltage step is applied at  Equivalent problems occur whenever the voltage is stepped, as it is in a typical voltage clamp experiment, or ramped rapidly, as is done in many single channel experiments, or if problems arise because of other physical effects, like series resistance in the bathing solutions.)

These irrelevant charging transients are taken care of in experiments (i.e., forced to a rapid uninteresting time scale) as much as possible, by the combination of the shunt capacitance and the mechanism maintaining the Dirichlet boundary condition on voltage (i.e., the voltage clamp). The equations should deal with the irrelevant charging transients the same way they are dealt with in experiments, both to be realistic, and also so we can recognize and deal with the artifacts using the wisdom of 60 years of experimentation on voltage clamp (since ([Hodgkin 1992](#_ENREF_15); [Huxley 1992](#_ENREF_22); [Huxley 2002](#_ENREF_23); [Hodgkin et al. 1952](#_ENREF_21)) and on single channels ([Sakmann and Neher 1995](#_ENREF_33); [Sigworth and Neher 1980](#_ENREF_37); [Hamill et al. 1981](#_ENREF_14); [Sigworth 1986](#_ENREF_36); [Sigworth 1995](#_ENREF_35); [Rae et al. 1988](#_ENREF_32); [Levis and Rae 1992](#_ENREF_25); [Rae and Levis 1992b](#_ENREF_30), [a](#_ENREF_29); [Levis and Rae 1993](#_ENREF_26); [Levis and Rae 1995](#_ENREF_27); [Tang et al. 1995](#_ENREF_38); [Levis and Rae 1998](#_ENREF_28); [Cherny et al. 2003](#_ENREF_3); [Rae and Levis 2004](#_ENREF_31))).

**Shunt Conductance.** The mathematical system must also be well posed, as it is in experiments, on the long time scale. The problems here come from diffusion forces and flows are very much weaker and act typically on a time scale of minutes or longer. Of course, integrated to time infinity, the diffusion forces and flows can produce concentration changes of great (even unlimited) importance. That is another of the issues we deal with here. Introducing the shunt conductance (or shunt resistance, typically 1011 ohm, i.e., 100 Gohm, or 10-11 Siemens, i.e., 10 pS), sometimes called the leakage conductance or the ‘membrane’ conductance (referring to the specific origin of the conductance when a lipid membrane is used to support an ion channel).

**Capacitance to ground, a physical low pass filter.** Finally, I introduce a new idea, that undoubtedly will have unforeseen consequences and ‘errors’ in application, because I have not had a chance to discuss it with colleagues, unlike the issues just discussed.

I suggest that we also include ***explicitly*** the capacitance to ground between bathing solutions (and channel, although that is not important) and the ground plane always placed just under the baths in real experiments. This capacitance to ground acts as a ‘low pass filter’; it dampens voltage changes by providing a path for capacitive current (proportional to  to leave the system. This damping decreases noise and transients in experiments making recordings much ‘cleaner’. I suspect — but do ***not*** know for sure – that including this capacitance to ground will make numerical problems easier. We will not know for sure until we implement the idea.

**Theoretical Model.** The equations we consider are just the PNP equations themselves, with ions treated as points, since as far as I can tell the issues of well posedness are the same in PNP as in more realistic models like the primitive model mentioned above.

 The central theme is that **electrical variables always take precedence. Equations should always integrate electrical variables explicitly**, dealing with chemical variables secondarily. Net charge should never be computed from  Indeed, the only use of that equation that I can justify is to compute the concentration of a species needed to make the solution electrically neutral, with to chemical precision. Chemical precision here means that  in chemical units is much smaller than any  to chemical precision, does not mean that the charge term on right hand side of the Poisson equation is zero!

This them implies a change in the numerical procedure and computer codes as usually written. **I believe one should explicitly write and integrate explicitly the flow equation for electrical current.**

Historically, the PNP equations as nearly always written to specify the flux of ions, not the flux of electrical charge. The current is the sum of the (weighted) fluxes of the PNP equation, plus the capacity current. The numerical implementation of PNP done this way is always on the edge of disaster. It is very easy somewhere to try to estimate NET charge , or flux of current, or charge on the boundary etc by summing the concentrations, flux of ions, or concentrations on the boundary.

***Any attempt to estimate electrical variables by summing ‘chemical’ variables is likely to produce disaster.***

We shall see that more is involved than simple algebra. There is a tension between the electrical boundary conditions I advocate and usual formulation for PNP, which specify all of the concentrations  corresponding to the fluxes , and not the current Specifically, the usual boundary conditions specify the  as constants imposed at either  or  They specify a potential applied across the system of *V* imposed at  and switched in a schedule to other potentials at other times. (These are the pulse patterns introduced to such useful effect by ([Hodgkin and Huxley 1952c](#_ENREF_18), [a](#_ENREF_16)) that form the basic paradigm of electrophysiology, ever since ([Bezanilla et al. 1982](#_ENREF_2)).

**Now, to Equations**

Here is a version of PNP in which Maxwell’s version of the continuity equation ***for charge*** is always used so the resulting flux equation for the flux  of charge (i.e., flux of total electric current) is always explicitly displayed and integrated.

Everything starts with Poisson

 

 

It is vital to remember that eq. can NOT be used to compute  because roughly speaking

  depending on location and conditions, always very small

The enormous strength of the electric field forces the net charge to be very small. If the net charge is not small, large gradients of potential exist which destroy biological systems. As a rule of thumb, membranes, proteins and ion channels are destroyed if potentials of 200 mV are applied to a system or fields are larger than 200 mV/(thickness of a lipid membrane = 2 nm)= 2x106 volts/cm. When  is this small, it cannot be computed by executing the sum shown in eq.. Indeed, eq. can be used to determine a particular concentration  from the other concentrations, if for some reason  is not known, but it can never be used to determine . Round off and other errors make eq. useless for computation. Rather, the Poisson equation itself must be used to compute .

 The flow equation for the total charge in the system without shunt capacitance is

 

or if we suppress Maxwell’s dreadful displacement field  (introduced before the electron was discovered and so before permanent charge was known to exist) and use more modern notation

 

This is eq. 6.4 of Jackson, p. 238, expanded, see p. 154. I follow Jackson’s notation exactly and use  for the permittivity (that has units. It is not the dielectric ‘constant’).

 Now we introduce the Nernst Planck equations for the flux of individual ions (NOT the flux of ionic current , rather the flux of the number of ions).

 

I use the diffusion coefficient and not mobility to avoid the confusion between the two definitions of mobility (absolute and electrical) in the literature. Obviously if the Einstein approximation fails and enough information is available to distinguish mobility from diffusion coefficient, the mobility should be used explicitly, and the choice of definition should be made explicitly.

It is important to remember that ions of different charge (i.e., valence) *zi* are very important in biology (calcium ions Ca2+ have charge +2; chloride Cl− ions have charge −1; sodium Na+ and potassium K+ ions have charge +1). They all have significantly different diffusion coefficients. It is ***NEVER*** permissible to treat all ions as having the same diffusion coefficient or the same (magnitude let alone sign) of charge if one wants to deal with biology.

Next, we relate eq. and by explicitly summing over all the ions to get total flux of charge

 

Maxwell Continuity eq. gives the equation for continuity of electric charge, i.e., current

 

I leave the permittivity inside the brackets so we never forget the assumption that is involved in moving it outside! And then we have the classical continuity equation for the flux of charge  (remember  is the flux of charge, not the flux of ions).

 

 

It is very important that no attempt be made to compute the sum on the right hand sides of eq.  from the concentrations . The sum  is ill posed, and must be nearly equal to zero, because of the enormous strength of electrical forces, summarized by the (approximate) physical principle of electroneutrality. One must never compute the sum  or its equivalent. Rather, one must compute  the net charge and its time derivative directly, for example, from the Poisson equation , without summing concentrations or even their rate of change.

In fact, the numerical properties of  are so different, depending on how it is computed, that it would be wise to use separate symbols in computer programs for the  of eq. and the  of eq. , on the right hand side of the Poisson equation. That way they will not be inadvertently interchanged

Note the form of the continuity of mass for each species

 

and the form for the continuity equation of total ‘mass’.

 

The mass continuity equations are ***not*** equivalent to the Maxwell continuity equation. They do not involve the charge  at all For example, the sum  is well posed and easy to compute because all its terms are positive while the corresponding terms in the sum for net charge  in eq.  can be positive or negative. In order to avoid confusion we do not even have a symbol for the flux of total mass .

**Boundary Conditions**.

We need to set boundary conditions on the electrical potential and charge, and write equations defining how we measure current  to correspond with experiment. Here is where we will introduce the shunt conductance and the shunt capacitance discussed in the beginning of this paper. The basic plan is to ***relax our boundary conditions on concentration and charge. We must allow one of the concentrations *** *(for example)* ***to ‘float’, i.e., to be determined by the rest of the problem. We also must allow the charge  to float, to be determined by the rest of the system. We do this by including additional pathways for flux and current. These additional pathways always exist in experiments. The pathway for shunt electrical current is the shunt capacitance . The pathway for individual ionic flux current is the leak or shunt conductance ***

The shunt capacitance drives a current ****** that adds to the current  of eq. once units are sorted out. The units of ****** are amps. The units of  are farads, i.e., Coulombs volt-1. The units of  in eq. are not the same as the units of ******. The current  in eq. must be scaled to be a new variable to deal with the geometry of the system. The units of the new (unnamed) scaled variable is amps, so it can be added to ****** in amps. The exact formulas depend on the geometry and so have to be worked out separately for each application, easily enough.

**Shunt Conductance.** The usual formulation of PNP sets boundary conditions on all *N* of the ionic species and ignores the experimental reality that  is measured, not  It is obviously impossible to specify *N* concentrations and deal with their fluxes as if they are independent, and also place conditions on the total current 

This apparent paradoxical situation is resolved when we realize that to make the system well posed (i.e., to reach steady state when concentrations and potentials on the boundaries are constants independent of time), we must introduce a shunt path that allows one of the concentrations to ‘float’, i.e., to take on any value as a function of time. The floating concentration is specified at  but not at later times.

In experiments, the leak flux ****** arises in many places. It can be flux through the lipid membrane, or through the seal between membrane and the plastic of the bath. Or it can be leakage through inadvertent unspecified pathways (i.e., dirt) that can easily arise in systems where resistances are of the order of 1011 ohms. (The resistance of most insulators is much less than 1011 ohms and so very special methods are necessary to record single channel currents ([Sakmann and Neher 1995](#_ENREF_33); [Sigworth and Neher 1980](#_ENREF_37); [Hamill et al. 1981](#_ENREF_14); [Sigworth 1986](#_ENREF_36); [Sigworth 1995](#_ENREF_35); [Rae et al. 1988](#_ENREF_32); [Levis and Rae 1992](#_ENREF_25); [Rae and Levis 1992b](#_ENREF_30), [a](#_ENREF_29); [Levis and Rae 1993](#_ENREF_26); [Levis and Rae 1995](#_ENREF_27); [Tang et al. 1995](#_ENREF_38); [Levis and Rae 1998](#_ENREF_28); [Cherny et al. 2003](#_ENREF_3); [Rae and Levis 2004](#_ENREF_31)). So we do not have a unique way to describe ****.**

I suggest that we describe it in the same ‘space’ of physics that we have been using., which, by the way, is the approach Hodgkin and Huxley used in their treatment of their leakage conductance ([Hodgkin and Huxley 1952e](#_ENREF_20), [b](#_ENREF_17); [Hodgkin and Huxley 1952d](#_ENREF_19), [c](#_ENREF_18), [a](#_ENREF_16); [Hodgkin et al. 1952](#_ENREF_21)).

I suggest we ***define*** the ****** by the conductance of the ion *m* of least interest, that changes the concentration in the bath most slowly. We add an additional shunt path ***for just this ion*** giving us now *N + 1* fluxes ****** where we only had *N* before. (I do not count the capacitive flux because its equation is so different in form and behavior.)

 

We choose values of  so the flux ****** has no significant effect on the properties of interest. ****** exists so that the concentration  can float (after it starts at an initial specified value). That way, we have *N* total flux equations. *N* ***−*** *1* of these are equations for individual ion fluxes, with Dirichlet boundary conditions at all times. One of these equations is the equation for total current ******, see eq. . The system is well posed.

 We now have two flux equations for the same ionic specie , one the real one and the other the leak. This does not cause trouble in the cases of interest because we can choose parameters for the leak path so the system is not perturbed in the time domain we study. The leak ensures that at infinite time the system will be stable.

 There is quite legitimate concern that not every system can be handled this way. For example, this approach will work if the flux of species  through ****** is not of interest, is small compared to the fluxes through the membrane, perhaps including ******, etc. It will work best if the baths are so large that the flux of  through ****** does not change concentrations we care about in the times of interest. A looser way to choose the properties of the leak is to choose parameters so the leak flux is 0.001 times the flux of the rest of the system. Typical values in experiments would be 1011 ohms (converting to strictly electrical units, for the entire system).

 In general, this approach will fail when experiments cannot be done. Such cases exist mathematically, and in thought experiments, but obviously not in the lab very often, so they have not been studied very much!!

**Shunt Capacitance.** Now, we introduce the shunt capacitance. We also must allow the charge  to float, to be determined by the rest of the system. We do this by including additional pathways for current. The additional pathway always exist in experiments. The pathway for shunt electrical current is the shunt capacitance

The shunt capacitance drives a current ****** that adds to the current  of eq. , once units are sorted out. The units of ****** are amps. The units of  are farads, i.e., Coulombs volt-1. The units of  in eq. are not the same as the units of ******. The current  in eq. must be scaled to be a new variable to deal with the geometry of the system. The units of the new (unnamed) scaled variable is amps, so it can be added to ****** in amps. The exact formulas depend on the geometry and so have to be worked out separately for each application, easily enough.

*The capacitance to ground remains. It is not dealt with explicitly in this document. Keep tuned!*

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