Why are voltage gated Na channels faster than K channels?

One multi-scale hierarchical model

Luigi Catacuzzeno¹, Luigi Sforna¹, Fabio Franciolini¹ and Robert S. Eisenberg²,³

¹ Department of Chemistry, Biology and Biotechnology, University of Perugia, Via Elce di Sotto 8, 06123 Perugia, Italy
² Department of Physiology and Biophysics, Rush University, Chicago IL 60612 USA
³ Department of Applied Mathematics, Illinois Institute of Technology, Chicago IL 60616 USA

*Correspondence to: Luigi Catacuzzeno
Department of Chemistry, Biology and Biotechnology
University of Perugia, via Elce di Sotto 8, 06123 Perugia - Italy
Phone number: +39-075-585-5755
Email address: luigi.catacuzzeno@unipg.it

Abstract

The generation of action potentials in excitable cells requires different activation kinetics of voltage gated Na (Nav) and K (Kv) channels. Nav channels activate much faster and allow the initial Na influx that generates the depolarizing phase and propagates the signal. Recent experimental results suggest that the molecular basis for this kinetic difference is an amino acid residue located in the gating pore of the voltage sensor domain, which is a highly conserved isoleucine in Kv channels, but an equally highly conserved threonine in Nav channels. Mutagenesis suggests that the hydrophobicity of this residue in Shaker Kv channels may regulate the energetic barrier that gating charges need to overcome to move through the gating pore, and ultimately the rate of channel opening. We use a multi-scale modeling approach to test this hypothesis. We use high resolution molecular dynamics to study the effect of the mutation on polarization charge within the gating pore. We then incorporate these results in a lower resolution model of voltage gating to predict the effect of the mutation on the movement of gating charges. The predictions of our hierarchical model is fully consistent with the tested hypothesis, thus suggesting that the faster activation kinetics of voltage gated Na channels comes from a stronger dielectric polarization by threonine (Nav channel) produced as the first gating charge enters the gating pore compared to isoleucine (Kv channel).
Introduction

Perhaps the most important role of voltage-gated sodium (Nav) and potassium (Kv) channels is their ability to generate action potentials, all-or-none (i.e., binary) rapid and transient changes of the electric potential difference across the plasma-membrane, that propagate signals long distances in the nervous system. This ability is essentially due to the different timing with which these two channels respond to a potential change (Figure 1A): Nav channels activate rapidly upon an initial depolarization of the membrane, allowing Na to enter the cell and induce a further membrane depolarization along the nerve during the first millisecond or so after the initial trigger. By contrast Kv channels remain essentially closed during this initial interval, and open only later, causing the repolarization of the membrane to the resting condition (Bezanilla, Rojas, and Taylor 1970; Hodgkin and Huxley 1952; Rojas, Bezanilla, and Taylor 1970).

The molecular and physical basis for the different kinetics of Nav and Kv channels have long been sought. The recently determined crystal structure of the voltage sensor domain (VSD) of Kv and Nav channels (Long et al. 2007; Long, Campbell, and MacKinnon 2005; Pan et al. 2018; Payandeh et al. 2011) has allowed investigation of the structure-function relationship of the voltage sensor, its environment, and the forces controlling its movement. In both channels the VSD is formed by the transmembrane segments 1-4 (S1-S4), with the S4 segment containing positively charged residues every third amino acid, and forming the voltage sensor, the structure that senses the electric field applied across the membrane and moves upon depolarization.

The S4 segment is surrounded for most of its length by wide intracellular and extracellular vestibules where water and ions can easily access and balance the gating charges (Figure 1B). The two vestibules are separated by a series of mostly hydrophobic residues that form a region, the gating pore, thought to be hardly accessible to water (green and red residues in Figure 1B). Several mutagenesis experiments suggest that some of these residues are in direct contact with the moving gating charges, and modulate the voltage-dependent properties of the channels (Lacroix et al. 2012, 2013, 2014; Lacroix and Bezanilla 2011; Tao et al. 2010).

One of the residues of the gating pore that determine the slow kinetics of Kv channels, compared to faster Nav channels, is isoleucine at position 287, located in the middle of the S2 segment (red in Figure 1B; notice that in this paper we will always use the residue numbering of the Shaker channel). This non polar, highly hydrophobic residue is found in many Kv channels. It is conserved by evolution presumably because of its essential role. By contrast, in the S2 segment of the first three domains of fast Nav channels, at corresponding position of isoleucine in Kv channels we consistently find the uncharged/polar hydrophilic residue threonine (Figure 1C) (Lacroix et al. 2013). Notably, when isoleucine at position 287 is mutated to the polar hydrophilic threonine (or to other polar amino acids), the Shaker gating currents become much faster and similar to the activation rate of Nav channels gating currents (Lacroix et al. 2013, 2014). These results strongly suggest that the physical structure and the underlying mechanism controlling the activation rate of voltage-gated ion channels involve the hydrophobic strength of the gating pore, especially of the residue at position 287. They suggest that a highly hydrophobic gating pore creates an

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1 We are grateful to F. Bezanilla for teaching us the importance of these results in a discussion that motivated the models of this paper.
energetically unfavorable condition, a high energetic barrier for the gating charges to move through upon depolarization, resulting in a slowing of the voltage sensor activation (Figure 1D). In this view, reducing the hydrophobicity of this region by replacing a highly hydrophobic residue with a more polar amino acid increases the polarizability of the region, lowers the energy barrier, provides counter charge (in the wall of the gating pore) for the gating charges, and ultimately increases the activation rate of the voltage sensor.

**Figure 1.** A) Schematic plot showing the activity of Nav and Kv channels during an action potential (red line). Nav channels open more rapidly than Kv channels, creating the depolarizing phase of the action potential. B) 3D structure of the active state of the VSD of the Shaker channel from (Henrion et al. 2012). The S4 segment is in blue and the S1-S3 segments are in cyan. The hydrophobic residues forming the gating pore (V236, I237, S240, I241, F244, C286, I287, F290, A319, I320; (Lacroix et al. 2014)) are shown in van der Waals (VdW) representation (green). The residue 287 is colored in red. Gating charge sidechains are shown in licorice representation and colored in yellow. C) Portion of the S2 segments of several voltage sensors belonging to Shaker, Kv and fast Nav channels, showing the highly conserved isoleucine at position 287 in Shaker-like potassium channels, and the threonine residue at the corresponding position in the fast Nav1.4 channel. D) Schematic representation of the hypothesis tested in this paper. Left: the S4 voltage sensor is represented as a cylinder with the gating charges in red. The rest of the VSD is in gray, and forms the gating pore and the intra- and extracellular vestibules. Right: energetic profiles encountered by the first gating charge across the gating pore, in presence of two different hydrophobicity levels of the gating pore. When the gating charge enters the gating pore there is an increase in the energy of the system, due to the destabilizing (hydrophobic) environment within the gating pore. The more hydrophobic is the gating pore, the higher will be this energy barrier and slower the activation rate.

We test this mechanistic hypothesis using a multi-scale modeling approach; that is, an approach that investigates the system at two different time and space scales, in a series of hierarchical computational methods whereby the calculated quantities from a computational simulation at an atomistic scale are used to define the parameters of the model at a coarser, macroscopic scale (Eisenberg 2018). In the specific case studied here, of voltage dependent gating, the idea translates into using Brownian modeling to macroscopically describe the system and obtain gating
currents, and then using molecular dynamics (MD) simulations to determine the needed free parameters present in the Brownian model. It is difficult to determine these parameters experimentally but there is hope from the exciting work of the Boxer group (Fried, Bagchi, and Boxer 2014; Fried and Boxer 2017; Suydam 2006). The multiscale approach appears feasible because assessing free parameters through MD simulations is much less susceptible to artifact and is less computationally demanding than simulating the overall movement of the voltage sensor.

Results and Discussion

The microscopic scale description
MD simulations performed on available crystal structures can be used to predict the polarizability of a protein or part of it, and thus its dielectric constant (Guest, Cashman, and Plotkin 2011; King, Lee, and Warshel 1991; Patargias, Harris, and Harding 2010; Simonson and Perahia 1996; Smith et al. 1993; Voges and Karshikoff 1998; Warshel et al. 2006), a parameter needed in the higher-scale Brownian modeling of voltage gating because it determines the amount of polarization charge available to balance movements of permanent charges, like those of the gating particle. We thus used MD to assess whether the substitution of threonine (found in the Nav channel) in place of isoleucine (found in the Kv channel) at position 287 can sensibly change the hydrophobic level and consequently the dielectric constant $\varepsilon$ of the gating pore, and ultimately the stability of the gating charges inside it.

Water accessibility estimates with MD
As a first step we determined the water accessibility of the VSD in the voltage-gated Shaker K channel by performing a MD simulation using the activated state of the Shaker VSD model. After inserting the protein in a phospholipid bilayer and in a water box, and performing the minimization, annealing, and equilibration of the overall system, we ran a 20 ns MD simulation to look for water inside the vestibules and the gating pore. At the end of the simulation we looked at the position of water molecules inside the VSD, to determine whether water could also access the gating pore.

The structure on the left of Figure 2A shows a frame at the end of a 20 ns MD simulation, while the one on the right shows the water-accessible space of the VSD obtained by the superposition of the water molecules from 30 consecutive frames (one every 2 ps). These pictures show that water readily enters both intracellular and extracellular vestibules.

The pictures also identify two distinct regions inside the gating pore with different water accessibility. One region, in direct contact with the intracellular vestibule, is totally inaccessible to water, thus forming an effective water inaccessible (WI) region. This region is very narrow – about 2-3 Å, as judged by the distance between the closest water molecules in the two vestibules (Figure 2D). It is located at the level of the gating pore in adjacency to residues F290 on S2, and V236 and I237 on S1 (Figure 2E). The second region, located between the WI region and the extracellular vestibule, including the other seven hydrophobic residues forming the gating pore (including
residue I287, Figure 2E), is water accessible (WA). As shown in Figure 2B, the equilibrated simulation shows about two water molecules in this region (when defined with a height of 6 Å above F290).

These results show the gating pore has two regions with different accessibility to water: a WI region totally inaccessible to water, and an adjacent WA region containing two water molecules most of the time. Due to the different water accessibility, we expect these two regions to have a quite different polarizability, with a relative dielectric constant close to 4 (the value typical for the...
interior of proteins) for the WI region, but sensibly higher for the WA region, since here water dipoles could significantly contribute to the stabilization of charges.

Estimating the dielectric constant of WA and WI regions of the gating pore using MD simulations

The dielectric constant has two main contributions, one given by the polarization of the electronic clouds, \( \varepsilon_{el} \), and the other due to the reorientation of permanent dipoles along the applied electric field, \( \varepsilon_{pd} \). Of these, \( \varepsilon_{el} \) can be measured experimentally (in bulk solutions) by impedance spectroscopy (Barsoukov and Macdonald 2018; Macdonald 1992) or by the refractive index of the material (Gussoni, Rui, and Zerbi 1998) that depends strictly on the dielectric constant at the appropriate frequency. In addition, the \( \varepsilon_{el} \) of a protein region may simply be crudely estimated by summing up the individual polarizability levels of the various amino acids present, that can be obtained experimentally (Voges and Karshikoff 1998). There is instead no experimentally recognized method to assess \( \varepsilon_{pd} \) inside a protein region, although it is known to depend on dipoles reorienting along the electric field. Indeed, the assignment of polarizability in an atomic scale system has serious ambiguities, as described in (Purcell and Morin 2012). For a small spherical volume of material surrounded by an infinite and homogeneous medium, \( \varepsilon_{pd} \) can be estimated by running extended MD simulations and analyzing the equilibrium fluctuation of the total dipole moment of the sample volume, by applying the Kirkwood-Frolich theory of dielectrics (Frohlich 1949; Simonson and Perahia 1996). Unfortunately, the structure we are studying – the voltage sensor – is immersed in a highly inhomogeneous medium (water and protein residues), on which this approach becomes impractical. In this case the dielectric constant in the small sample volume in the protein interior will also depend on the polarizability of the surrounding medium, whose value is neither known and surely cannot be assumed to be homogeneous. The assumption of homogeneity is crucial because inhomogeneities create dielectric boundary charges that can easily dominate the properties of a system (Nadler et al. 2004; Nadler, Hollerbach, and Eisenberg 2003).

We thus reasoned that under our conditions the dielectric constant of the sample volume could be estimated with a different approach: from MD simulations run at variable applied electric fields. More specifically, using the basic definition of polarization (P) as the sum of the dipole moments (\( \mu \)) per unit volume (Vol), and its relation to the dielectric constant (\( \varepsilon_{pd} \)), usually valid at moderately low electric fields, we can write:

\[
P = \frac{\Sigma \mu_i}{Vol} = \varepsilon_0 (\varepsilon_{pd} - 1)E
\]

where \( \varepsilon_0 \) is the permittivity of vacuum, E is the applied electric field, and Vol is the sample volume. The total dipole moment of the sample can be determined using time averaging over an MD simulation (<\( M > \)) while applying an electric field along the z direction (E\(_z\)), and this allows us to calculate \( \varepsilon_{pd} \) as

\[
\varepsilon_{pd} = 1 + \frac{< M >_z}{E_z Vol \varepsilon_0}
\]
Figure 3. A) Running average of the dipole moment assessed for a water molecule inside a 30 Å cubic water box, during three MD simulations performed under different applied electric fields along the z direction. B) Plot of the mean total dipole moment of a water molecule as a function of the applied electric field. Data obtained at applied electric fields of ±0.2 kcal/(mol·Å·e) were fitted with a straight line, and the dielectric constant reported in the plot was estimated from the slope obtained. Notice that 1 kcal/(mol·Å·e) corresponds to about 0.4 Volts/nm. C) Plots of the mean total dipole moment at two different applied electric fields, for the water accessible region (top) and for the water inaccessible region (bottom) of the gating pore, assessed from MD simulations using the wild type (WT (I287), left) of the mutated (MUT (I287T), right) structure of the VSD. From these data the $\varepsilon_{pd}$ was estimated and reported in each plot. The $\varepsilon_{el}$ was estimated from the Clausium-Mosotti relationship $\varepsilon_{el}^{-1} \varepsilon_{el}^{-2} = \frac{\alpha}{3 \varepsilon_0 \text{Vol}}$, where $\alpha$ is the polarizability and Vol is the volume of the region considered, estimated from the molar polarizability of the amino acidic residues reported in Table 1 of (Voges and Karshikoff 1998). D) 3D structures of the gating pore in the WT and mutated structures. The gating pore residues are shown in VdW representation, in green or orange/blue depending on their location in the WA or WI region of the gating pore, respectively. Residues 287 and 240, located very close to the gating charge, are colored in blue. Also water molecules inside the WA region are shown in VdW representation (red/white). The gating charge R371 is shown in licorice representation (yellow). E) Plot of the mean dielectric constant in the WA and WI regions of the gating pore, estimated as $\varepsilon = \varepsilon_{el} + \varepsilon_{pd}$ as shown in panel C, on six different MD simulations (three for the WT and three for the mutant structures).

This method emerges naturally from consideration of the relation of charge and electric fields in proteins (Eisenberg 1996) and has been successfully applied to estimate the dielectric constant of a protein, as well as of water confined inside the pore of a synthetic ion channel (Sansom et al.
In order to validate this method for estimating \( \varepsilon_{pd} \), we first applied it to water molecules in a cubic water box, that is known to have a dielectric constant close to 82 for our TIP3P water model (Simonson 1996). We ran several MD simulations under different electric fields, applied along the \( z \) direction, and measured each time the \( z \) component of the total dipole moment associated to one water molecule (chosen at random) inside the cubic box.

The plot in Figure 3A reports a running average of the total dipole moment assessed for three different applied electric fields, and shows that the value of the dipole moment varies with the applied fields. Figure 3B illustrates the relationship between the total dipole moment along the \( z \) direction and the applied electric field. As expected this relationship is linear for moderate applied fields, while it saturates at higher values. From the slope of this relationship within the linear regime we obtained a dielectric constant of about 86, a value quite close to what expected, indicating that our method to estimate the dielectric constant works satisfactorily in this situation.

We then applied this method to estimate the dielectric constant of the WI and WA regions of the gating pore, both in the WT and in the mutant structure in which isoleucine 287 (found in Kv channels) had been replaced by threonine (mutant I287T), found in Nav channels. In the calculation of the dielectric constant of the WI region we considered all three residues present in this region, namely residues F290, V236 and I237, since they all surround the gating charge while it moves through this region of the gating pore (Figure 3D, green residues). As expected for a very hydrophobic region, the polarizability of the WI region was electronic in nature, with a value around 4.0, irrespective of the residue present at position 287 (Figure 3C, E).

The situation is quite different for the WA region where the gating pore widens, and not all residues are close to the moving gating charge. In the assessment of the dielectric constant we thus considered only those residues adjacent to the gating charge (i.e. placed at a distance less than 4 Å), namely, residues I287 and S240, plus two or three water molecules (Figure 3D, blue residues and red/white waters). The dielectric constant estimated in this region was much higher than in the WI region, but, most relevant, it was quite different in the two WA structures considered, being about 20 in the WT and about 60 in I287T mutant (Figure 3C, E). These results indicate that the nature of the residue at position 287 critically contributes to the polarizability of the WA region around the gating charge, and a mutation of hydrophobic isoleucine at this position with a more hydrophilic amino acid like threonine markedly increases the polarization charge in this region.

Two main reasons explain why the WA region has a higher polarizability in the mutant compared to WT. First, threonine residue has a much higher permanent dipole, because of its hydroxyl group, and thus stronger hydrophilic power than isoleucine. As a result, the dipole moment contributed by this one residue is much higher in the I287T mutant than in the WT structure (Figure 4A). Second, the WA region in the I287T mutant appears to have a slightly higher water content, with a number of water molecules of about 1.8 for WT and 2.2 for I287T (Figure 4B), presumably because threonine stabilizes water molecules inside this region, perhaps with interactions that form a weak hydrogen bond, more or less.
Figure 4. A) Running average of the dipole moment associated to residue 287 assessed for the WT and the mutated structure, during two MD simulations performed with a different applied electric field along the z direction. The mutant threonine is much more polarizable than WT isoleucine. B) Running average of the water molecules found inside the WA region of the gating pore in MD simulations performed using either the WT or the I287T mutated structure of the VSD.

The macroscopic scale description

The macroscopic scale description of voltage gating is based on the Brownian model that we have recently developed, and shown to well predict the major signature features of experimental gating currents of Shaker channels (Catacuzzeno and Franciolini 2019; Catacuzzeno, Sforna, and Franciolini 2020) (see Methods). Other papers (Horng et al. 2019; Kim and Warshel 2014; Peyser et al. 2014; Peyser and Nonner 2012b, 2012a) give other views of the voltage sensor and resulting currents at different resolutions. Only (Horng et al. 2019) and (Catacuzzeno and Franciolini 2019) actually calculate gating currents comparable to those measured experimentally under a range of conditions, with (Horng et al. 2019) having lower atomic detail. Our model takes the relevant structural and electrostatic parameters directly from a 3D structural model of the Shaker VSD (Figure 5A and B). The electro-diffusion of ions in the bath is described through a flux conservation equation that assumes a quasi-equilibrium, on the time scale of gating currents, and the motion of S4 segments is described by Brownian dynamics. The electric potential profile is modeled using the Poisson equation, and includes all the charges present in the system, including the gating charges on the S4 segment and the fixed charges on segments S1-S3 of the VSD. Notably the Poisson equation is used in our model to assess the electric potential profile contains the relative dielectric constant, ε, as an undetermined parameter that we have estimated above for both the WT and mutant gating pores, using MD simulations. It should clearly be understood that this form of coarse graining has the significant advantage that the lower resolution treatment satisfies conservation laws of mass, charge, and current. Coarse graining procedures that do not explicitly introduce the Poisson equation and boundary conditions are unlikely to satisfy conservation laws of charge and current.

In our previous model we assumed a 2 Å long gating pore localized at the level of F290, and that separated electrically the intracellular and extracellular solutions. The dielectric constant was set to 4 in this short region, and to 80 elsewhere, including the adjacent region containing residue I287 (Catacuzzeno and Franciolini 2019). The gating pore used in the present study has been changed to account for our MD results shown above. More specifically, the gating pore is now
pictured as two distinct regions: one with all the properties previously described, with 2 Å length comprising F290, and fully impermeant to water (green bar in Figure 5A and B).

**Figure 5.** A) Schematics showing the geometry of the VSD assumed in our model. The S4 segment containing the gating charges was assumed to move perpendicular to the membrane through the gating pore (7 Å long) and the extracellular and intracellular vestibules (each 31.5 Å long, and opening with a half angle of 15°). Notice that, compared to the model presented in (Catacuzzeno and Franciolini 2019), the gating pore is 5 Å longer to accommodate several hydrophobic residues including residue I287. The vestibules are correspondingly shorter to give the same overall length. The dashed lines represent some of the surfaces delimiting the volume elements considered in our numerical simulations (see (Catacuzzeno and Franciolini 2019) for details). B) Profile of the gating pore radius illustrating the 2 Å region, comprising F290 and assumed to be fully impermeant to water (WI, green bar), and the water accessible 5 Å region comprising I287 (WA, red bar), the dielectric constant (ε), the fixed charge density located in the S1-S3 region of the VSD (Z_F), and the charge density on the S4 segment (Z_S4), for the Shaker model structure. x=0 was placed in the center of the gating pore. C) Plot showing the total energy associated to the S4 segment, assessed as \( G_{el} = \frac{1}{2} k_B T \langle x - x_{eq} \rangle^2 + \int Z_{S4} \varepsilon_0 \frac{dV(x)}{dx} dx \), as a function of the S4 segment position. The electrostatic energy profile displays five clear energy wells associated to five different positions of the voltage sensor (represented in schematic form). Inset: Simulated macroscopic gating current evoked by a voltage pulse from -90 to -20 mV, using our Brownian model of the voltage sensor.
The second region, 5 Å long, comprising several hydrophobic residues, including I287, is accessible to water, and has now been assigned a dielectric constant of 20, in accordance with the MD results (Figure 5A and B). As shown in Figure 5C the updated model continues to predict the major features of experimental gating currents obtained in response to a depolarizing pulse. First, a very fast gating current component is present at the beginning of the depolarizing step, rising instantaneously and then falling very rapidly.

Figure 6. A) Profile of the gating pore radius and the dielectric constant along the VSD of the WT and the I287T mutated models of voltage gating. The two models only differ in the dielectric constant in the WA region of the gating pore. B) Simulated gating currents obtained in response to depolarizing pulses from a holding potential of -90 mV for both the WT (black) and the mutated (red) models. C) Plot of the decay time constant of the ON gating currents shown in panel B as a function of voltage. When the gating current decay was clearly bi-exponential, a mean of the two time constants weighted for the respective areas was reported. Inset: superimposed ON gating currents obtained in response to a depolarization to -20 mV for the two models. D) Plot of the mean charge vs the applied voltage, obtained by integrating the ON gating currents over 40 ms long simulations at different applied voltages. No appreciable difference was observed in the two models. E) Simulated gating currents obtained in response to repolarizing pulses after a short (4 ms) depolarizing step to 20 mV, for the WT (black) and the mutated (red) models. F) Plot of the decay time constant of the OFF gating currents shown in panel E as a function of voltage. Inset: superimposed OFF gating currents obtained in response to a repolarization to -80 mV for the two models.

Second, a later slower current component develops, starting with a plateau/rising phase and proceeding with a slow decay. Both these features have been observed in experiments (Bezanilla 2000; Sigg, Bezanilla, and Stefani 2003). It is comforting that the fast component appears without
adding any features to the model. The plot in Figure 5C represents the total energy profile assessed for the various allowed positions of the voltage sensor. It can be seen that the model predicts five energy wells corresponding to five different positions of the S4 segment, agreeing with the view of many papers that the voltage sensor moves in multiple steps during the activation process (Bezanilla, Perozo, and Stefani 1994; Henrion et al. 2012; McCormack, Joiner, and Heinemann 1994; Schoppa et al. 1992; Tytgat and Hess 1992; Zagotta, Hoshi, and Aldrich 1994). In our model each step represents the passage of a different gating charge through the high resistance and water inaccessible region of the gating pore (cf. drawings in Figure 5C).

**Testing the effects of a dielectric constant change on the gating current kinetics**

We verified that a different dielectric constant of the WA region can explain the different kinetics of the macroscopic gating current of WT and I287T mutant found experimentally (Lacroix et al. 2013). We used our macroscopic Brownian model of voltage gating to see whether changing the dielectric constant of the WA region from 20 to 60 reproduces the increased rate of gating current seen in mutagenesis experiments.

As shown in Figure 6B and C there was a significant increase in the rate of change of the gating currents evoked by depolarizing pulses, especially at intermediate levels of depolarization (Figure 6C), as observed experimentally. The model also predicted an increased rate of the OFF gating currents induced by repolarization after a short depolarizing pulse (Figure 6E and F). No change was instead observed in the charge vs voltage relationship (Figure 5D). Overall, these results are very similar to those observed experimentally with the mutation I287T, suggesting that a change in the dielectric constant within the gating pore is a major determinant of the functional effects induced by the mutation.

We then examined the effects of changing the polarizability of the WA region on the electrostatic energy profile, and found that the mutation selectively lowers the first of the four peaks present in the profile (Figure 7). This result appears quite reasonable since the first energetic peak essentially represents the dielectric boundary energy commonly (but incorrectly\(^2\)) called the Born energy for the addition of one (gating) charge to the WA region of the gating pore (Nadler et al. 2004, 2003), and we expect this quantity to depend on the dielectric constant present there (cf. inset to Figure 7). The remaining energy peaks are instead essentially unmodified as the voltage sensor moves, presumably because the charge residing in the WA region is simply replaced by the next charge of the S4 segment, as the voltage sensor moves in response to an applied voltage, keeping the overall charge content in the WA region constant.

\(^2\) The Born energy was computed by Born in an infinite homogeneous domain without dielectric discontinuities. The system we study is about as far from homogeneous as one could imagine and its biological functions are determined by the heterogeneous distribution of polarization and dielectric properties. The use of the phrase Born energy is seriously misleading in the opinion of at least one of the authors.
Figure 7. Plot showing the electrostatic energy associated with the S4 segment, assessed as $G_{el} = \int Z_{S4} e_0 \frac{dV(x)}{dx} dx$, as a function of the S4 segment position, for the two models of voltage-dependent gating differing by the dielectric constant in the WA region of the gating pore. Inset: schematic drawing showing that the change in the electrostatic energy exclusively at the first peak may be explained by the addition of the first gating charge to the WA region of the gating pore.

Conclusions

Several conclusions may be drawn from the results of our multiscale analysis. First, using MD simulations we have shown that the gating pore of the Shaker channel, defined in terms of the ten residues proposed by (Lacroix et al. 2014), does not have uniform accessibility to water. More specifically, we have found that the region totally inaccessible to water is actually very short, having a length of 2-2.5 Å and including the three most intracellular residues, namely F290 on the S2 segment, and V236 and I237 on S1. The remaining part of the gating pore, covering a length of about 5 Å, is by contrast accessible to water, with two water molecules often found in this region, as shown by our simulations. The potential difference applied to the membrane will then drop mostly over the very short WI region, because this part of the gating pore is the only region with very high resistance to water and ions. The notion of a very narrow high-resistance region has already been suggested by several studies (Ahern and Horn 2005; Asamoah et al. 2003; Islas and Sigworth 2001).
A second result of this study is the quantitative demonstration that the polarizability (i.e., the hydrophobicity) of the residue at position 287 of Shaker channels can greatly modulate the field strength acting on the gating charge as it enters the gating pore. We showed this by estimating the dielectric constant of the region surrounding the gating charge, using MD simulations at variable applied electric fields, and assessing the change in the total dipole moment of the region. In contrast to the computational methods usually employed to assess the dielectric constant (Guest et al. 2011; King et al. 1991; Simonson and Perahia 1996; Smith et al. 1993; Warshel et al. 2006), that are based on the Kirkwood-Frolich theory of spatially uniform dielectrics (Frohlich 1949; Simonson and Perahia 1996), our method does not assume a homogeneous environment of known polarizability surrounding the sample volume of interest. For this reason our approach can be applied also to restricted regions inside a protein, containing a small number of residues. This method has been already successfully applied to estimate the dielectric constant of a homogeneous liquid, as well as of water confined inside the pore of a synthetic ion channel (Kolafa and Viererblová 2014; Riniker, Kunz, and Van Gunsteren 2011; Sansom et al. 1997; Xu et al. 1996; Yeh and Berkowitz 1999). However, to the best of our knowledge, it has not been applied before to assess the local dielectric constant inside a (channel) protein.

We found that changing the residue at position 287 significantly changes the dielectric constant acting on the gating charge. More specifically, when the isoleucine found at this position in Shaker (as well as in other Kv channels) is changed to threonine, the residue found at the corresponding position in Nav channels, the estimated dielectric constant appears to increase by as much as three times. The observed increase may have two components: i) the much higher permanent dipole present on the threonine side-chain (as compared to isoleucine, mostly because of the hydroxyl group there), that contributes directly to increase the polarizability, and ii) the possibility that threonine, as a polarizable dipole, attracts more waters inside the WA region of the gating pore, electrostatically and by establishing weak hydrogen bonds (Perlstein 2001). In summary, our data show that the WA region of Nav channel has a higher dielectric constant, and this possibly allows the first gating charge to enter its gating pore more easily than in Kv channels. The change in dielectric constant lowers the energetic barrier gating charge encounters in entering the gating pore.

We also used our Brownian model (Catacuzzeno and Franciolini 2019) to see if the predicted change in the dielectric constant found by MD explains the functional effect of the I287T mutation found in experiments. Experiments show a noticeable increase in both the ON and OFF gating currents, but only a small change in the charge-voltage relationship. The output of our Brownian model was quite similar to the experimental results, suggesting that the polarizability change caused by the I287T mutation is a major determinant for the effects observed on the macroscopic gating currents. We then looked at the mechanism linking the dielectric constant of the WA region to the rate of movement of the voltage sensor by analyzing the effects of the mutation on the energy profile encountered by the voltage sensor as it moves into the gating pore. We found that the reduction in the dielectric constant of the WA region produces a selective decrease in the first of the four energy peaks encountered by the voltage sensor, corresponding to the entry of the first gating charge (R1) into the gating pore. This result appears quite reasonable since this step involves the entry of a charge into the WA region, a process modulated by its
polarizability. It is important to realize that what we show is a prediction of gating currents performed without using adjustable parameters, not a fitting routine where free parameters are adjusted to match the model output with the experimental data (Horng et al. 2019).

In this study we have analyzed in detail the functional properties of Shaker channel gating, with isoleucine or threonine at location 287, since the type of residue present at this position (or equivalent position in Nav channels) is thought to create the kinetic differences between Nav and Kv channels (Lacroix et al. 2013). The results obtained suggest that different activation kinetics of voltage-gated Na and K channels comes from different dielectric forces exerted by threonine compared to isoleucine, as the first gating charge enters the WA region of the gating pore. It seems that evolution uses a single side chain to control the speed of the Kv and Nav channels, and ultimately to allow the action potential to exist.

Other processes could contribute to the time course of Na and K channels, as pointed out by Bezanilla’s group. First, residue 363 in S4 is a highly conserved threonine in domains I, II and III of Nav channels, while is a highly conserved isoleucine in Kv channels. Mutation of this isoleucine to threonine makes the voltage sensor of Kv channels move faster, as inferred from the time course of the gating currents (Lacroix et al. 2013). From the position of this residue relatively far from the gating charge R1, we suspect that its contribution to the polarizability would not be significant, suggesting that a different mechanism is likely responsible for this effect. Second, the presence of the β1 subunit of Nav channels also accelerates the voltage sensor, with unknown mechanism (Lacroix et al. 2013). Finally, the sensors in Nav channels exhibit positive cooperativity which also accelerates the overall speed of channel activation (Chanda, Asamoah, and Bezanilla 2004). We point out that (within the limitations of our models) what we propose is both logically sufficient and logically necessary to account for the difference in speed and the structural movements of the voltage sensor. It is sufficient because we reproduce the properties of gating current under a realistic range of conditions. It is necessary because the movement of charges that calculate must produce the currents we calculate given the universal and exact nature of the Maxwell equations that link charge and current. Of course, evolution is not logical. It often provides redundant mechanisms that are beyond the necessary and (perhaps barely) sufficient. These mechanisms may provide properties not glimpsed in gating currents studied with step functions in the way they are usually measured. These mechanisms may make the gating system robust in conditions we have not explored.

A similar multiscale approach may also be used in the future to understand the functional effects of the many other mutations that have been performed on the VSD, and so to have a physical interpretation of the data and a more advanced understanding of the voltage-dependent gating.

**Methods**

**Molecular dynamics simulations.** The voltage sensor domain used in the simulation was taken from the Shaker model of Henrion et al. (2012), in the activated state. For MD simulations, the protein was embedded on a membrane environment made of POPC, with dimensions of 80x80 Å, and solvated in two steps. The distance between the maximum and minimum z coordinates and the water box edges in the z-axis was set to 15 Å. The system was first
solvated below the membrane plane with a water box of dimensions 83.46, 80.05, 16.19 (x, y, z in Å) where 9462 TIP3 water molecules (Jorgensen et al. 1983) were placed. Then the system was solvated above the membrane plane with a water box of the dimensions 83.46, 80.05, 20.02 (x, y, z in Å) where a total of 9462 TIP3 water molecules were placed. After the addition of the water molecules, the system presented a total net charge of -2. The system was neutralized and a total of 27 Cl ions plus 29 Na ions were added to the system making up a salt concentration of 0.15 mol/L. The MD simulations in the present study were performed employing the NAMD molecular dynamics package (Phillips et al. 2005). The CHARMM36 force field (Best et al. 2012; MacKerell et al. 1998) was used in all MD simulations. A distance cut-off of 12.0 Å was applied to short-range, non-bonded interactions, and of 10.0 Å for the smothering functions. Long-range electrostatic interactions were treated using the particle-mesh Ewald (PME) (Darden, York, and Pedersen 1993) method. Before the MD simulations all the systems were submitted to an energy minimization protocol for 2000 steps. After minimization an annealing of 0.29 ns was performed, with a temperature ramp of 0.24 ns of simulation where the temperature was raised from 60 K to 300 K. The pressure was maintained at 1 atm using Nosé-Hoover Langevin piston (Feller et al. 1995; Martyna, Tobias, and Klein 1994). The equations of motion were integrated using the r-RESPA multiple time step scheme (Phillips et al. 2005) to update the short-range interactions every 1 step and long-range electrostatics interactions every 2 steps. The time step of integration was chosen to be 2 fs for all simulations. During the Annealing the motion of the backbone of the protein was restrained. After the annealing a 10 ns equilibration was performed, while maintaining the restriction of the backbone motion, during which the temperature was maintained at 300 K using Langevin dynamics. The pressure was maintained at 1 atm using Nosé-Hoover Langevin piston. A distance cut-off of 12.0 Å was applied to short-range, non-bonded interactions, and 10.0 Å for the smothering functions. Long-range electrostatic interactions were treated using the particle-mesh Ewald (PME) method. Finally, we performed unrestricted MD simulations by applying different values of the electric field in the z direction. These simulations were used to assess the running average of total dipole moment, used to estimate the dielectric constant as described in the “Results” section.

The Brownian model of voltage gating. A full description of the Brownian model of voltage gating can be found in (Catacuzzeno and Franciolini 2019; Catacuzzeno et al. 2020). Briefly, in our model the VSD was approximated to a hourglass-shaped geometrical structure consisting of a cylindrical gating pore, having a length of 7 Å and a diameter of 10 Å, flanked by internal and external water accessible vestibules having a length of 3.15 nm each and a conical shape opening with a half angle of 15° into two hemispherical subdomains of bath solution, both having radius of 1 μm. The S4 charge profile (Z_{S4}) was built by considering the six positive charges, whose mean distance between the charged atoms was determined from a 3D Shaker channel model, and each giving rise to a charge profile normally distributed with a standard deviation of 0.1 nm. The fixed charged profile (Z_{F}) was similarly built by considering the vertical dimension of the Cα of all the charged residues on segments S1-S3 of the VSD.

Ions moved by electro-diffusion governed by the following flux conservation equation:

\[
\frac{dC_j(x,t)}{dt} = -\nabla F_j(x,t)
\]

where \(C_j(x,t)\) is the concentration of ion \(j\), \(t\) is the time, \(\nabla\) is the spatial gradient operator, and \(F_j(x,t)\) is the flux (mole per second per unit area) of ion \(j\), given by the Nernst-Planck equation:

\[
F_j(x,t) = -D_j(x) \left[ \nabla C_j(x,t) + \frac{z_jF}{RT} \nabla V(x,t) \right]
\]

where \(D_j(x)\) and \(z_j\) are the diffusion coefficient profile and the valence of ion \(j\), respectively, \(F\), \(R\) and \(T\) have their usual meanings, and \(V(x,t)\) is the electrical voltage profile. Because ions move much faster in the bath than the voltage sensor moves in the gating a quasi-equilibrium assumption was made for ions electrodiffusion by imposing \(F_j(x,t) = 0\).

The S4 segment was assumed to move in one dimension as a Brownian particle, whose dynamics is governed by the following Langevin’s equation

\[
x_{S4}(t) = \frac{F_{ex}(x_{S4}, t)}{m} + r(t)
\]

Here \(x_{S4}(t)\) represents the position of the voltage sensor (distance between the R2-R3 midpoint and the center of the gating pore), \(m\) is the mass of the particle, \(F_{ex}(x_{S4}, t)\) is the external force acting on the particle, and \(R(t)\) is a random force due to the collision of the fluid and the rest of the protein on the S4 segment, which has a probability distribution with zero mean and second moment given by \(<R(t)R(t')>=2k_BT\delta(t-t')\) where \(k_B\) is the
Boltzmann constant and \( \delta \) is the delta function. The external force was assumed to have an electrical component, assessed from the electric voltage profile, and a spring component with force constant of 0.005 N/m to account for the force exerted by the channel pore, which in this model has not been considered (Catacuzzeno and Franciolini 2019). \( \gamma \), the friction coefficient of the S4 voltage-sensor, was set to \( 2 \times 10^{-6} \) Kg/s based on the comparison between experimental and predicted macroscopic gating currents (notice that in Catacuzzeno et al., 2019 this parameter was set to twice this value).

The dynamics of the S4 segment may also be described in terms of the time evolution of the probability density function profile, given by the following Fokker-Planck (FP) equation:

\[
\frac{df_{S4}(x,t)}{dt} = -\frac{\partial}{\partial x} \left( \mu(x) f_{S4}(x,t) \right) + \frac{k_B T}{\gamma} \frac{\partial^2 f_{S4}(x,t)}{\partial x^2}
\]

Here \( f_{S4} \) represents the probability density function for the position of the S4 segment and \( \mu(x) \) is \( F_{ex}(x_{S4}, t)/\gamma \). This partial differential equation was solved with elastic boundary conditions at \( x = \pm 1.8 \) nm in order to set the allowed movement of the particle to 3.6 nm. The elastic boundary condition was imposed by setting the particle flux equal to zero at the boundaries:

\[
\text{Flux} = \left( \mu f - \frac{k_B T \delta f(x,t)}{\gamma} \right)_{x=\pm 1.8 \text{ nm}} = 0
\]

Once determined, the dynamics of the probability density function \( f_{S4}(x,t) \) was used to estimate the macroscopic gating currents, to be compared with those obtained experimentally. However, since the solution of the FP equation required the steady-state approximation for the electrolyte ions dynamics (cf. above), the gating current could not be directly obtained from eqn. (2), but was assessed by analyzing the net charge changes in the left (or alternatively in the right) bath, according to the following equation (Horn et al. 2019):

\[
I_{g}(t) = \int_{x_{S4}-1.8 \text{ nm}}^{x_{S4}+1.8 \text{ nm}} (dQ_{\text{bath}}(x) / dx) dx \\
\]

Where \( Q_{\text{bath}}(x_{S4}) = e_0 \int_{\text{left bath}} A(x)(\sum z_i C_i) dx = e_0 \int_{\text{right bath}} A(x)(\sum z_i C_i) dx \) represents the net charge found in the left (or right) bath when the position of the S4 segment is \( x_{S4} \) and the integration covers the whole left (right) bath. We checked that the total current density was independent of location \( x \) at all times even though the components of the current vary a great deal with \( x \) (Eisenberg 2019, 2020; Horn et al. 2019) and of course time \( t \).

The electrical voltage profile \( V(x) \) was assessed from the net charge density profile \( \rho(x) \), using the following Poisson’s equation

\[
\varepsilon_0 \left( \frac{d}{dx} \left( \varepsilon(x) \frac{dV(x)}{dx} \right) + \varepsilon(x) \frac{dV(x)}{dx} \cdot \frac{dA(x)}{dx} \right) = -\rho(x)
\]

Where \( \varepsilon_0 = 8.854 \times 10^{-12} \text{ F-m}^{-1} \) is the vacuum permittivity, \( \varepsilon(x) \) is the position dependent dielectric coefficient, \( V(x) \) is the electric potential and \( A(x) \) is the position dependent surface. Programs were written in C and are available upon request. Figures were prepared using Origin v4.0 and CorelDraw v18.1.

References


